Subject Code: B132103/R13

II B. Pharmacy I Semester Regular Examinations Dec/Jan. – 2014-15 PHYSICAL PHARMACY-II

Time: 3 hours Max. Marks: 70

Question Paper Consists of **Part-A** and **Part-B**Answering the question in **Part-A** is Compulsory,
Three Questions should be answered from **Part-B*******

PART-A

- 1. (i) Define and differentiate diffusion and osmosis
 - (ii) Define rate and order of the reaction with simple examples
 - (iii) Define and differentiate surface tension and interfacial tension?
 - (iv) Why do we need to reduce the particle size of a drug and what are the advantages associated with it?
 - (v) What is the main difference between Newtonian and non-Newtonian systems
 - (vi) Define dispersion, dispersed phase and dispersion medium with a suitable examples

[4+4+4+4+3+3]

PART-B

- 2. (a) Define solubility and classify solubility according to I.P.
 - (b) Discuss in detail about factors influencing the solubility of solids in liquid

[6+10]

3. Write a note on decomposition pathways of medicinal agents and strategies adopted for their stabilization in the pharmaceutical formulations.

[16]

4. Describe the concept of HLB system along with its applications and limitations

[16]

5. What are derived properties of powders and how they are useful in the design and development of various pharmaceutical formulations?

[16]

6. What is thixotropy and mention its applications in the design of pharmaceutical formulations?

[16]

7. Write in detail about stability of colloids?

[16]

